

This article was downloaded by:

On: 22 January 2011

Access details: *Access Details: Free Access*

Publisher *Taylor & Francis*

Informa Ltd Registered in England and Wales Registered Number: 1072954 Registered office: Mortimer House, 37-41 Mortimer Street, London W1T 3JH, UK



Journal of Asian Natural Products Research

Publication details, including instructions for authors and subscription information:

<http://www.informaworld.com/smpp/title~content=t713454007>

Two new secoiridoid glycosides from the leaves of *Olea europaea* L.

Xiao-Fei Wang^{ab}; Chen Li^a; Yan-Ping Shi^a; Duo-Long Di^a

^a Key Laboratory for Natural Medicine of Gansu Province, Lanzhou Institute of Chemical Physics, Chinese Academy of Sciences, Lanzhou, China ^b The Graduate University of Chinese Academy of Sciences, Chinese Academy of Sciences, Beijing, China

To cite this Article Wang, Xiao-Fei, Li, Chen, Shi, Yan-Ping and Di, Duo-Long (2009) 'Two new secoiridoid glycosides from the leaves of *Olea europaea* L.', *Journal of Asian Natural Products Research*, 11: 11, 940 – 944

To link to this Article: DOI: 10.1080/10286020903310979

URL: <http://dx.doi.org/10.1080/10286020903310979>

PLEASE SCROLL DOWN FOR ARTICLE

Full terms and conditions of use: <http://www.informaworld.com/terms-and-conditions-of-access.pdf>

This article may be used for research, teaching and private study purposes. Any substantial or systematic reproduction, re-distribution, re-selling, loan or sub-licensing, systematic supply or distribution in any form to anyone is expressly forbidden.

The publisher does not give any warranty express or implied or make any representation that the contents will be complete or accurate or up to date. The accuracy of any instructions, formulae and drug doses should be independently verified with primary sources. The publisher shall not be liable for any loss, actions, claims, proceedings, demand or costs or damages whatsoever or howsoever caused arising directly or indirectly in connection with or arising out of the use of this material.

Two new secoiridoid glycosides from the leaves of *Olea europaea* L.

Xiao-Fei Wang^{ab}, Chen Li^a, Yan-Ping Shi^{a*} and Duo-Long Di^{a*}

^aKey Laboratory for Natural Medicine of Gansu Province, Lanzhou Institute of Chemical Physics, Chinese Academy of Sciences, Lanzhou 730000, China; ^bThe Graduate University of Chinese Academy of Sciences, Chinese Academy of Sciences, Beijing 100039, China

(Received 14 May 2009; final version received 4 September 2009)

Two new secoiridoid glycosides, oleuricines A (**1**) and B (**2**), together with five known triterpenoids, β -amyrin, oleanolic acid, erythrodiol, urs-2 β ,3 β -dihydroxy-12-en-28-oic acid, and β -maslinic acid, were isolated from the EtOAc-soluble part of EtOH extract of the leaves of *Olea europaea* L. The structures of these compounds were elucidated by various spectroscopic methods, including intensive 1D, 2D NMR, and HR-ESI-MS techniques.

Keywords: *Olea europaea* L.; secoiridoid glycosides; oleuricine A; oleuricine B

1. Introduction

Olea europaea L. has been widely cultivated in the Mediterranean area. The leaves of *O. europaea* L. have been used as a folk medicine by native people for the treatment of fever and other diseases, such as malaria. *O. europaea* L. has been transplanted into Sichuan and Gansu Provinces of China since last century. Secoiridoid glycosides are the most important chemical constituents of the genus *Olea* [1,2] and the biological activities of these compounds, such as antimicrobial, anti-inflammatory, hypotensive, and hypoglycemic activities, have been reported previously [3–5]. As a continuing investigation of bioactive constituents of the leaves of *O. europaea* L., two new secoiridoid glycosides, together with five known triterpenoids, β -amyrin, oleanolic acid, erythrodiol, urs-2 β ,3 β -dihydroxy-12-en-28-oic acid, and β -maslinic acid, were isolated (Figure 1). Here, the

isolation and structural elucidation of the new compounds are reported.

2. Results and discussion

Compound **1** was obtained as a yellow gum. Its molecular formula was established as C₃₁H₄₂O₁₈ on the basis of HR-ESI-MS at m/z 720.2719 [M+NH₄]⁺. The IR spectrum showed the absorption bands for hydroxyl (3374 cm⁻¹), α , β -unsaturated carbonyl (1629 and 1702 cm⁻¹), and aromatic ring (1520 cm⁻¹). The structure of compound **1** could be elucidated on the basis of 1D and 2D NMR spectral data as well as comparison with the literature [6]. The ¹³C NMR spectrum of **1** showed some signals for substructures of a secoiridoid, a hydroxytyrosol, and a β -glucopyranosyl, which were similar to that of the known oleuropein [6]. However, there was a set of additional signals (Table 1) due to a monosaccharide moiety, which was determined as a β -D-glucopyranosyl group by

*Corresponding authors. Email: didl@lzb.ac.cn; shiyp@lzb.ac.cn

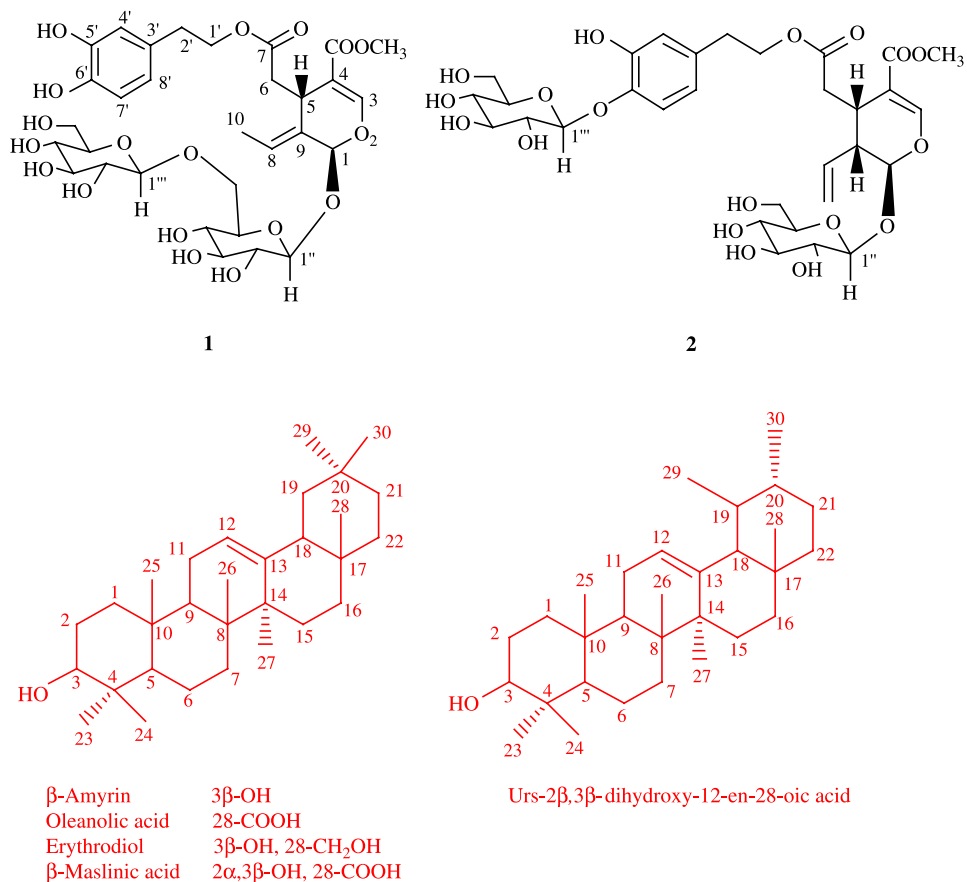


Figure 1. Chemical structures of isolated compounds.

comparing with that of methyl- β -D-glucopyranoside [7]. The result agreed with the acid hydrolysis assay. In this experimentation, D-glucopyranose was identified by paper chromatography (PC) through comparison with authentic sugars. Both D-glucopyranose moieties could be confirmed as the β -configuration on the basis of the anomeric proton signals at δ 4.19 (1H, d, $J = 7.6$ Hz) and 4.63 (1H, d, $J = 8.0$ Hz) in the ¹H NMR spectrum. The linkage between glucosyl groups was determined by the shift effects compared to oleuropein: C-6'' at downfield δ 69.9 and C-5'' at upfield δ 73.3 in **1**. This was confirmed by the key correlation between H-1''' and C-6'' in the HMBC experiment (Figure 2). Thus, the structure of **1** was

determined to be 6''-O- β -D-glucopyranosyl-oleuropein, named as oleuricine A.

Compound **2** was obtained as a yellow gum. Its molecular formula C₃₁H₄₂O₁₈ was deduced from HR-ESI-MS at m/z 725.2251 [M+Na]⁺. The IR spectrum showed the absorption bands for hydroxyl (3361 cm⁻¹), α,β -unsaturated carbonyl (1629 and 1703 cm⁻¹), and aromatic ring (1509 cm⁻¹). The ¹³C NMR spectrum of **2** was similar to that of oleuroside [6], except for the presence of another β -D-glucopyranosyl group (Table 1), which was deduced by comparing with methyl- β -D-glucopyranoside. This was confirmed by the acid hydrolysis assay. Acid hydrolysis of compound **2** gave D-glucopyranose, which was identified by PC comparison

Table 1. ^1H NMR (400 MHz) and ^{13}C NMR (100 MHz) spectral data of **1** and **2** in $\text{DMSO}-d_6$ (TMS, δ in ppm, J in Hz).

No.	1		2	
	δ_{H}	δ_{C}	δ_{H}	δ_{C}
Secoiridoid				
1	5.85 (s)	92.9	5.38 (d, 4.4)	95.4
3	7.50 (s)	153.4	7.44 (s)	152.1
4		107.8		108.1
5	3.85 (dd, 4.0, 8.0)	30.1	3.11–3.19	27.6
6	2.37–2.43 (m)	39.9	2.65 (m)	33.8
7		170.7		171.7
8	5.94 (m)	123.1	5.49 (m)	133.2
9		129.3	2.72 (m)	43.2
10	1.66 (d, 6.0)	13.1	5.18 (m)	119.5
11		166.2		166.5
OMe	3.60 (s)	51.3	3.58	51.1
Hydroxytyrosol				
1'-a	4.01 (m)	64.1	4.08–4.15 (m)	64.6
1'-b	4.25 (m)		4.08–4.15 (m)	
2'	2.64 (m)	35.1	2.66 (m)	34.0
3'		129.3		129.7
4'	6.58 (s)	116.3	6.68 (s)	117.1
5'		145.0		144.1
6'		143.5		146.9
7'	6.59 (d, 8.0)	115.5	7.00 (d, 8.0)	116.5
8'	6.44 (d, 8.0)	119.5	6.58 (d, 8.4)	120.0
Glc				
1''	4.63 (d, 8.0)	99.0	4.48 (d, 8.0)	98.6
2''	3.35 (m) ^a	73.6	3.25 (m) ^a	73.3
3''	3.35 (m) ^a	76.4	3.24 (m) ^a	75.9
4''	3.07 (m) ^a	70.0	3.12 (m) ^a	70.0
5''	2.97 (m) ^a	73.3	3.11 (m) ^a	77.2
6''	3.31 (m) ^a	69.9	3.67 (m) ^a	61.1
Glc				
1'''	4.19 (d, 7.6)	102.9	4.58 (d, 7.2)	102.8
2'''	3.07 (m) ^a	73.3	2.95 (m) ^a	73.0
3'''	3.07 (m) ^a	76.5	3.10 (m) ^a	76.7
4'''	3.56 (m) ^a	70.0	3.02 (m) ^a	69.8
5'''	3.16 (m) ^a	77.4	3.07 (m) ^a	77.4
6'''	3.39 (m) ^a	61.2	3.34 (m) ^a	60.8

^aSignal pattern unclear due to overlapping.

with authentic sugars. In addition, the anomeric proton signals at δ 4.48 (1H, d, $J = 8.0$ Hz) and 4.58 (1H, d, $J = 7.2$ Hz) were recognized in the ^1H NMR spectrum. The coupling constants indicated that both the D-glucopyranose moieties were of β -configuration. The β -glucopyranosyl group was linked at C-6' of the hydroxytyrosol moiety, because of the C-6' signal at downfield δ 146.9 compared with signals of oleuroside. This was also

confirmed by the HMBC correlation between H-1''' and C-6' (Figure 2). Thus, the structure of compound **2** was determined to be 6'-O- β -D-glucopyranosyl oleuroside, named as oleuricine B.

3. Experimental

3.1 General experimental procedures

Optical rotations were recorded on a Rudolph-Research Autopol-III automatic

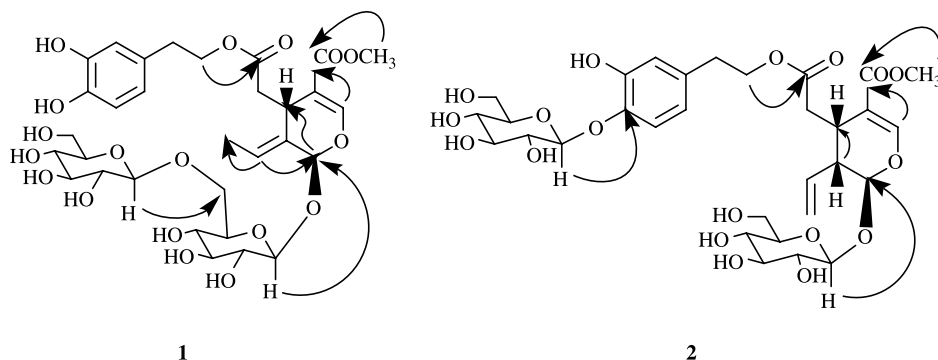


Figure 2. Key HMBC correlations of **1** and **2**.

polarimeter. The UV spectra were measured on a T6 UV-vis spectrophotometer (Beijing Purkinje General Instrument Co., Ltd, Beijing, China). The IR spectra were obtained on a Nicolet-170SX FT-IR spectrometer (Madison, WI, USA); KBr pellets in cm^{-1} . The HR-ESI-MS spectra were recorded on a Bruker APEX-II mass spectrometer in m/z . The ^1H and ^{13}C NMR spectra were recorded on a Varian Inova-400 FT-NMR spectrometer (Palo Alto, CA, USA), with TMS as an internal standard, δ in ppm, J in Hz. Silica gel (200–300 mesh) for column chromatography and silica gel GF₂₅₄ for TLC were purchased from Qingdao Marine Chemical Factory, Qingdao, China. C₁₈ reversed-phase (RP) silica gel (35–75 mm) was obtained from Analteck, Inc. (Newark, NJ, USA).

3.2 Plant material

The leaves of *O. europaea* L. were collected from Longnan County of Gansu Province in China on February 2008. The plant was identified by Prof. Huan-Yang Qi in Lanzhou Institute of Chemical Physics, Chinese Academy of Sciences, Lanzhou, China, and a voucher specimen (No. ZY200803) has been deposited at Key Laboratory for Natural Medicine of Gansu Province, Lanzhou Institute of Chemical Physics, Chinese Academy of Sciences, China.

3.3 Extraction and isolation

The air-dried leaves of *O. europaea* L. (2.5 kg) were extracted with EtOH (25 liters) at room temperature three times (each time for 5 days). The EtOH extract was evaporated under reduced pressure at 50°C to yield a residue (1 kg). The residue was suspended in H₂O (2000 ml) and then partitioned with petroleum ether (60–90°C, 6×1500 ml) and EtOAc (6×1500 ml) successively to give petroleum ether and EtOAc fractions. Then, the EtOAc fraction (300 g) was subjected to column chromatography eluted with a gradient of CHCl₃–CH₃OH. The fraction (100:5, v:v, 2500 mg) was separated to give four subfractions (A–D). On the basis of TLC analysis, subfraction A (600 mg) was further separated over a silica gel column eluting with petroleum ether–acetone (9:1) to yield β -amyrin (20 mg) and oleanolic acid (45 mg). Repeated silica gel column separation of subfraction D (800 mg) using the eluent petroleum ether–acetone (8:1) afforded erythrodiol (16 mg), urs-2 β ,3 β -dihydroxy-12-en-28-oic acid (25 mg), and β -maslinic acid (30 mg). The fraction (100:20, v:v, 500 mg) was chromatographed over silica gel column eluting with EtOAc–CH₃OH (13:1) to yield three subfractions (E–G) on the basis of TLC analysis. Subfraction F (100 mg) was resubjected to column chromatography (C₁₈ RP silica gel,

CH₃OH–H₂O 15:85) for several times to afford **1** (15 mg) and **2** (10 mg).

3.3.1 Oleuricine A (**1**)

A yellow gum, C₃₁H₄₂O₁₈; $[\alpha]_D^{20}$ –137 ($c = 1.00$, MeOH); UV (MeOH) λ_{\max} (log ϵ): 282 (6.55), 244 (6.77) nm; IR (KBr) ν_{\max} (cm⁻¹): 3374, 2922, 1702, 1629, 1521, 1439, 854; ¹H and ¹³C NMR spectral data: see Table 1; HR-ESI-MS: m/z 720.2719 [M+NH₄]⁺ (calcd for C₃₁H₄₆O₁₈N⁺, 720.2709).

3.3.2 Oleuricine B (**2**)

A yellow gum, C₃₁H₄₂O₁₈; $[\alpha]_D^{20}$ –54 ($c = 1.00$, MeOH); UV (MeOH) λ_{\max} (log ϵ): 276 (6.38), 248 (6.55), 234 (6.54) nm; IR (KBr) ν_{\max} (cm⁻¹): 3361, 2921, 1705, 1630, 1509, 1436, 823; ¹H and ¹³C NMR spectral data: see Table 1; and HR-ESI-MS: m/z 725.2251 [M+Na]⁺ (calcd for C₃₁H₄₂O₁₈Na⁺, 725.2263).

3.4 Acid hydrolysis

A solution of compound **1** or **2** (each c.5 mg) in H₂O (8 ml) was treated with 2 N HCl (2 ml) at 80°C for 10 h. The solution

was evaporated under reduced pressure to yield a residue. The residue was dissolved in MeOH. The solvent system EtOAc–pyridine–H₂O (2:2:1) was used for PC identification of glucose.

Acknowledgements

This work was financially supported by the ‘Hundred Talents Program’ of Chinese Academy of Sciences.

References

- [1] G. Montedoro, M. Servili, M. Baldioli, R. Selvaggini, E. Miniati, and A. Macchioni, *J. Agric. Food Chem.* **41**, 2228 (1993).
- [2] M. Brenes, A. Garcia, P. Garcia, J.J. Rios, and A. Garrido, *J. Agric. Food Chem.* **47**, 3535 (1999).
- [3] G. Bisignano, A. Tomaino, R.L. Cascio, G. Crisafi, N. Uccella, and A. Saija, *J. Pharm. Pharmacol.* **51**, 971 (1999).
- [4] C.M. Bitler, T.M. Viale, B. Damaj, and R. Crea, *J. Nutr.* **135**, 1475 (2005).
- [5] F. Visioli, G. Bellomo, C. Montedoro, and C. Galli, *Atherosclerosis* **117**, 25 (1995).
- [6] H. Kuwajima, T. Uemura, K. Takaishi, K. Inoue, and H. Inouyet, *Phytochemistry* **27**, 1757 (1988).
- [7] P.A.J. Gorin and M. Mazurek, *Can. J. Chem.* **53**, 1212 (1976).